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pressure) you have created a physical change. Examples of physical changes -

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mixture of solid and liquid. At point 'C' the state is liquid. At point 'D' the liquid is evaporating, it is a mixture of liquid and gas. At point 'E' the state is gas. Temperature 'X' is the melting point while temperature 'Y' is the boiling point.

IGCSE - Chemistry

Matter contains particles such as atoms, ions or molecules. The matter has three states, namely solid, liquid and gaseous. This division is based on the differences in physical properties such as mass, volume, shape, rigidity, density and arrangement of particles. The kinetic theory of matter states that all matter:

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Kornelsen 1 Matter is composed of small particles (molecules) 2 The molecules are in constant motion This motion is different for the three states of matter
Solid - Molecules will bend

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molecules are exactly the same (same ratio of elements, same geometric bonding pattern), but we encounter water in three different forms in the world around us.

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XI Chemistry | Unit: States of Matter: Gases And liquids | Lecture 4

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