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## **Electrolytes Worksheet Answer Key**

Worksheet 5 - Solutions, Electrolytes and Concentration The name electrolyte is given to substances whose aqueous solutions contain ions, because ions are charged particles which, when they move through a solution, conduct electricity. Strong electrolytes dissociate completely into ions. This includes the strong acids (HCl, HBr, HI, HNO<sub>3</sub>, HClO<sub>4</sub> and H<sub>2</sub>SO<sub>4</sub>)

## **Worksheet 5 - Solutions, Electrolytes and Concentration**

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This quiz and worksheet combo will help you quickly assess your understanding of solutions, solutes, mixtures and electrolytes. You will also be assessed on how to determine molarity and mass percent.

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## **Quiz & Worksheet - Solutions, Electrolytes and ...**

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10. Are the following strong, weak, or non-electrolytes? a.  $\text{PbSO}_4$  weak f.  $\text{CH}_3\text{OH}$  non b.  $\text{Al}_2(\text{SO}_4)_3$  strong g.  $\text{K}_2\text{CrO}_4$  strong c.  $\text{C}_6\text{H}_{12}\text{O}_6$  non h.  $\text{Ni}(\text{NO}_3)_2$  strong d.  $\text{SrI}_2$  strong i.  $\text{Ba}(\text{C}_2\text{H}_3\text{O}_2)_2$  strong e.  $\text{Cu}(\text{OH})_2$  weak j.  $\text{Ca}_3(\text{PO}_4)_2$  weak

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Answer the following questions based on the solubility curve below, Name 150 130 120 100 90 80 70 60 50 40 30 20 ...

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Chemistry IF8766 . ELECTROLYTES Name Electrolytes are substances that break up (dissociate or ionize) in water to produce ions, ...

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10. Which of the following is a non-electrolyte? Mg(OH) 2 B) H 2 SO 4 C) P 2 S 3 D) CaCl 2 11. What am I? a- I allow electric current to flow through water. b- When dissolved in water, I



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do not allow electric current to flow through it. \_\_\_\_ c- My electrolytic dissociation provides ions other than  $H^+$  and  $OH^-$  ions. \_\_\_\_ d- I am an ...

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Electrolyte, in chemistry and physics, substance that conducts electric current as a result of a dissociation into positively and negatively charged particles called ions, which migrate toward and ordinarily are discharged at the negative and positive terminals (cathode and anode) of an electric circuit, respectively.

## **electrolyte | Definition, Examples, & Facts | Britannica**

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