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Nuclear magnetic resonance (NMR)

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spectroscopy provides detailed information about the nature and concentration of many components in a chemical reaction because most molecules contain NMR-active nuclei. However, the application of NMR to in operando analysis under pressures greater than 1-2 atm has been limited by the difficulties of handling high

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pressures and achieving good gas-liquid mass transport in the NMR instrument.

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will build up pressure until it reaches equilibrium and exists as a liquid at room temperature (at approximately 10 atm of pressure).

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analyzing organic and inorganic compounds, and reaction kinetics. Until recently, it has only been possible to use NMR at relatively modest pressures since the modelling used to predict the various properties, cannot be verified at high pressures.

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